

# Dimensional analysis #1

Name KEY

**Show all set-ups!!**

Given the following information:

$1 \text{ quark} = 2.9 \text{ whos}$

$1 \text{ whos} = 5 \text{ mabees}$

$1 \text{ bug} = 3.7 \text{ quarks}$

$1 \text{ kuz} = 3.2 \text{ mabees}$

Change:

- 1.) 3.0 quark to bug  $\rightarrow 3.0 \text{ q} \times \frac{1 \text{ b}}{3.7 \text{ q}} = 0.81081081 \text{ b}$  1. 0.81 b  

$q \rightarrow b$
- 2.) 1.3 mabees to quarks  $\rightarrow 1.3 \text{ m} \times \frac{1 \text{ w}}{5 \text{ m}} \times \frac{1 \text{ q}}{2.9 \text{ w}} = 0.089655172 \text{ q}$  2. 0.090 q  

$m \rightarrow w \rightarrow q$
- 3.) 7.0 whos to bugs  $\rightarrow 7.0 \text{ w} \times \frac{1 \text{ q}}{2.9 \text{ w}} \times \frac{1 \text{ b}}{3.7 \text{ q}} = 0.652376514 \text{ b}$  3. 0.65 b  

$w \rightarrow q \rightarrow b$
- 4.) 8.75 kuz to quarks  $\rightarrow 8.75 \text{ k} \times \frac{3.2 \text{ m}}{1 \text{ k}} \times \frac{1 \text{ w}}{5 \text{ m}} \times \frac{1 \text{ q}}{2.9 \text{ w}} = 1.931034483 \text{ q}$  4. 1.93 q  

$k \rightarrow m \rightarrow w \rightarrow q$
- 5.) 2.5 bug to kuz  $\rightarrow 2.5 \text{ b} \times \frac{3.7 \text{ q}}{1 \text{ b}} \times \frac{2.9 \text{ w}}{1 \text{ q}} \times \frac{5 \text{ m}}{1 \text{ w}} \times \frac{1 \text{ k}}{3.2 \text{ m}} = 41.9140625 \text{ k}$  5. 42 k  

$b \rightarrow q \rightarrow w \rightarrow m \rightarrow k$
- 6.)  $2.0 \times 10^{-3}$  quarks to mabees  $\rightarrow 2.0 \times 10^{-3} \text{ q} \times \frac{2.9 \text{ w}}{1 \text{ q}} \times \frac{5 \text{ m}}{1 \text{ w}} =$  6. 0.029 m  

$q \rightarrow w \rightarrow m \rightarrow k$
- 7.) 900. quarks to whos  $\rightarrow 900 \text{ q} \times \frac{2.9 \text{ w}}{1 \text{ q}} = 2610 \text{ w}$  7. 2610 w  

$q \rightarrow w$
- 8.)  $4.8 \times 10^5$  kuz to mabees  $\rightarrow 4.8 \times 10^5 \text{ k} \times \frac{3.2 \text{ m}}{1 \text{ k}} = 1536000 \text{ m}$  8.  $1.5 \times 10^6 \text{ m}$   

$k \rightarrow m$
- 9.) 205 mabees to whos  $\rightarrow 205 \text{ m} \times \frac{1 \text{ w}}{5 \text{ m}} = 41 \text{ w}$  9. 41.0 w  

$m \rightarrow w$
- 10.) 15 bug<sup>2</sup> to quarks<sup>2</sup>  $\rightarrow 15 \text{ b}^2 \times \frac{3.7^2 \text{ q}^2}{1^2 \text{ b}^2} = 205.35 \text{ q}^2$  10. 210 q<sup>2</sup>  

$b^2 \rightarrow q^2$
- 11.) 2.5 mabees<sup>2</sup> to kuz<sup>2</sup>  $\rightarrow 2.5 \text{ m}^2 \times \frac{1^2 \text{ k}^2}{3.2^2 \text{ m}^2} = 0.24 \text{ k}^2$  11. 0.24 k<sup>2</sup>  

$m^2 \rightarrow k^2$
- 12.)  $1.50 \times 10^3$  kuz<sup>2</sup> to bug<sup>2</sup>  $\rightarrow$ 

$k^2 \rightarrow m^2 \rightarrow w^2 \rightarrow q^2 \rightarrow b^2$

12. 5.34 b<sup>2</sup>  
 $1500 \text{ k}^2 \times \frac{3.2^2 \text{ m}^2}{1^2 \text{ k}^2} \times \frac{1^2 \text{ w}^2}{5^2 \text{ m}^2} \times \frac{1^2 \text{ q}^2}{2.9^2 \text{ w}^2} \times \frac{1^2 \text{ b}^2}{3.7^2 \text{ q}^2} = 5.336441625 \text{ b}^2$

## DIMENSIONAL ANALYSIS

Set up and solve the following problems using dimensional analysis. Be sure to express your results to the proper number of significant figures.

- ✓1. How many seconds are there in 1.2 weeks?
- ✓2. How many centimeters are there in 4.38 feet?
- ✓3. How many meters did you run if the distance run was  $6.59 \times 10^5$  inches?
- ✓4. What is the mass of a suitcase, in pounds, if it weighs 19.5 kilograms?
- ✓5. If a recipe calls for 37 grams of sugar, how many pounds does that correspond to?
- ✓6. Express a volume of  $589 \text{ cm}^3$  in  $\text{ft}^3$  and  $\text{in}^3$ .
- ✓7. How many liters are equal to  $39 \text{ in}^3$ ?
- ✓8. If a car travels at  $4.45 \times 10^4 \text{ ft/hr}$ , what would its speed be in meters/min?
- ✓9. What is the density of a substance if it has a mass of 59.2 grams and a volume of 17.0 mL?
- ✓10. Calculate the density of a liquid, in  $\text{grams/cm}^3$ , if it has a mass of 23.2 grams and occupies a cube with dimensions of  $1.3 \text{ cm} \times 5.6 \text{ cm} \times 2.3 \text{ in}$ .
- ✓11. If a liquid has a density of 1.04 g/mL, what would its density be in  $\text{lb/in}^3$ ?
- ✓12. What volume, in liters, would 88.9 grams of a substance occupy if its density is 2.38 g/mL?
- ✓13. What is the mass, in pounds, of 389 mL of a gas that has a density of 1.29 g/L?
- ✓14. Convert  $37^\circ\text{C}$  to  $^\circ\text{F}$  and K.
- ✓15. Which temperature is the coldest?  
a)  $-12^\circ\text{C}$     b)  $18^\circ\text{F}$     c) 248K

$$1. \quad \frac{1.2 \text{ weeks} \left| \frac{7 \text{ days}}{1 \text{ wk}} \right| \frac{24 \text{ hr}}{1 \text{ day}} \left| \frac{60 \text{ min}}{1 \text{ hr}} \right| \frac{60 \text{ sec}}{1 \text{ min}} = 725760$$

there is an exact # of seconds

$$2. \quad \frac{4.38 \text{ ft} \left| \frac{12 \text{ in}}{1 \text{ ft}} \right| \frac{2.54 \text{ cm}}{1 \text{ in}} = 134 \text{ cm}$$

$$3. \quad 16,700 \text{ m}$$

$$4. \quad 43.0 \text{ lb}$$

$$5. \quad 0.082 \text{ lb}$$

$$6. \quad 35.9 \text{ in}^3, \quad 0.0208 \text{ ft}^3$$

$$7. \quad 0.64 \text{ L}$$

$$8. \quad 226 \text{ m/min}$$

$$9. \quad 3.49 \text{ g/mL}$$

$$10. \quad 23.2 \text{ g} / 43 \text{ cm}^3 = 0.55 \text{ g/cm}^3$$

$$11. \quad 0.0376 \text{ lb/in}^3$$

$$12. \quad 0.0374 \text{ L}$$

$$13. \quad 0.00111 \text{ lb}$$

$$14. \quad 99^\circ\text{F}, \quad 310 \text{ K}$$

$$15. \quad \text{a) } -12^\circ\text{C} \quad \text{b) } -7.8^\circ\text{C} \quad \text{c) } -25^\circ\text{C}$$

c) is coldest

# KEY

## Dimensional Analysis Practice Problems

1)  $0.56 \text{ kg} = ? \text{ mg}$

$$0.56 \text{ kg} \times \frac{1000 \text{ g}}{1 \text{ kg}} \times \frac{1 \text{ mg}}{0.001 \text{ g}} = \underline{560,000 \text{ mg}}$$

2)  $1.2 \text{ ng} = ? \text{ g}$

$$1.2 \text{ ng} \times \frac{10^{-9} \text{ g}}{1 \text{ ng}} = \underline{1.2 \times 10^{-9} \text{ g}}$$

3)  $2.0 \text{ in} = ? \text{ mm}$  ( $1 \text{ in} = 2.54 \text{ cm}$ )

$$2.0 \text{ in} \times \frac{2.54 \text{ cm}}{1 \text{ in}} \times \frac{0.01 \text{ m}}{1 \text{ cm}} \times \frac{1 \text{ mm}}{0.001 \text{ m}} = \underline{51 \text{ mm}}$$

4)  $500 \text{ ft} = ? \text{ m}$

$$500 \text{ ft} \times \frac{12 \text{ in}}{1 \text{ ft}} \times \frac{2.54 \text{ cm}}{1 \text{ in}} \times \frac{0.01 \text{ m}}{1 \text{ cm}} = \underline{152.4 \text{ m}} \rightarrow \underline{152 \text{ m}}$$

5)  $10 \mu\text{L} = ? \text{ cc}$  ( $1 \text{ mL} = 1 \text{ cm}^3 = 1 \text{ cc}$ )

$$10 \mu\text{L} \times \frac{10^{-6} \text{ L}}{1 \mu\text{L}} \times \frac{1 \text{ mL}}{0.001 \text{ L}} \times \frac{1 \text{ cc}}{1 \text{ mL}} = \underline{0.01 \text{ cc}}$$

6)  $3 \text{ wk} = ? \text{ min}$

$$3 \text{ wk} \times \frac{7 \text{ day}}{1 \text{ wk}} \times \frac{24 \text{ hr}}{1 \text{ day}} \times \frac{60 \text{ min}}{1 \text{ hr}} = \underline{30,240 \text{ min}}$$

7)  $50 \text{ mL} = ? \text{ cups}$  ( $1 \text{ L} = 4.226 \text{ cups}$ )

$$50 \text{ mL} \times \frac{0.001 \text{ L}}{1 \text{ mL}} \times \frac{4.226 \text{ cups}}{1 \text{ L}} = \underline{0.2113 \text{ cups}}$$

8)  $5.33 \text{ km} = ? \text{ dm}$

$$5.33 \text{ km} \times \frac{1000 \text{ m}}{1 \text{ km}} \times \frac{1 \text{ dm}}{0.1 \text{ m}} = \underline{53,300 \text{ dm}}$$

9)  $123.0 \text{ ng} = ? \text{ Mg}$

$$123.0 \text{ ng} \times \frac{10^{-9} \text{ g}}{1 \text{ ng}} \times \frac{1 \text{ Mg}}{10^6 \text{ g}} = \underline{1.23 \times 10^{-13} \text{ Mg}}$$

10)  $3 \text{ yds} = ? \text{ in}$  ( $1 \text{ yd} = 3 \text{ ft}$ )

$$3 \text{ yds} \times \frac{3 \text{ ft}}{1 \text{ yd}} \times \frac{12 \text{ in}}{1 \text{ ft}} = \underline{108 \text{ in}}$$

rounded

CHEM 160  
Worksheet: Dimensional Analysis KEY

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1.  $9 \text{ in} \rightarrow \text{cm}$

$$\cancel{9} \cancel{\text{in}} \times \frac{2.54 \text{ cm}}{1 \cancel{\text{in}}} = 9 \times 2.54 \text{ cm} = 22.86 \text{ cm} \rightarrow 20 \text{ cm}$$

1 sig. fig.

2.  $2916402 \rightarrow \text{kg}$

$$\cancel{29.25} \cancel{\text{lb}} \times \frac{0.45359 \text{ kg}}{1 \cancel{\text{lb}}} = 13.2675075 \text{ kg} \rightarrow 13.27 \text{ kg}$$

4 sig. fig.

3.  $210 \text{ cm} \times \frac{1 \text{ in}}{2.54 \text{ cm}} = 82.67 \text{ in} \rightarrow$

83 in  
2 sig. fig.

4.  $3.05 \cancel{\text{in}} \times \frac{39.37 \cancel{\text{in}}}{1 \cancel{\text{in}}} \times \frac{1 \text{ ft}}{12 \cancel{\text{in}}} = \frac{3.05 \times 39.37}{12} \text{ ft} = 10.03917 \text{ ft}$

10.0 ft  
3 sig. fig.

5.  $14 \text{ gal} \times \frac{3.785 \text{ L}}{1 \text{ gal}} = 52.99 \text{ L} \rightarrow$

53 L  
2 sig. fig.

6.  $2.5 \cancel{\text{L}} \times \frac{1 \text{ qt}}{0.94634 \cancel{\text{L}}} = 2.6 \text{ qt}$

2 sig. fig.

This is less than 3 qt so it's not enough.

7.  $20 \cancel{\text{hr}} \times \frac{2.47 \cancel{\text{hr}}}{1 \cancel{\text{hr}}} \times \frac{1 \text{ owl}}{3 \cancel{\text{hr}}} = 16.466667 \text{ owl} \rightarrow$

20 owl (1 sig. fig.)

8.  $8 \cancel{\text{d}} \times \frac{1 \cancel{\text{hr}}}{2.47 \cancel{\text{d}}} \times \frac{1 \text{ sheep}}{0.125 \cancel{\text{hr}}} = \frac{8}{2.47 \times 0.125} \text{ sheep} =$   
 $25.91093117 \text{ sheep} \rightarrow 30 \text{ sheep (1 sig fig)}$

9.  $1 \text{ pk} = 34 \text{ g carbs}$   
 $6 \text{ days} \times \frac{1 \text{ pk}}{\text{day}} \times \frac{34 \text{ g carbs}}{\text{pk}} \times \frac{16 \text{ oz}}{454 \text{ g}} = \frac{6 \times 34 \times 16}{454} \text{ oz carbs} =$   
 $7.189427313 \text{ oz carbs} = 7 \text{ oz carb (1 sig. fig.)}$

10.  $9 \text{ g fat} = 1 \text{ bar}$  ;  $1 \text{ pack} = 0.6 \text{ dbar}$  ; (a) ? oz fat (b) ? cal  
 $1 \text{ pack} \times \frac{0.6 \text{ dbar}}{1 \text{ pack}} \times \frac{10 \text{ bar}}{1 \text{ dbar}} \times \frac{9 \text{ g fat}}{1 \text{ bar}} \times \frac{1 \cancel{\text{kg}}}{453.6 \cancel{\text{g}}} \times \frac{16 \text{ oz}}{1 \cancel{\text{lb}}} =$   
 (a)  $\frac{0.6 \times 10 \times 9 \times 16}{453.6} \text{ oz fat} = 1.904761905 \text{ oz} \approx 2 \text{ oz fat}$   
 (b)  $0.6 \times 10 \times 9 \text{ g fat} \times \frac{9 \text{ cal}}{1 \cancel{\text{g fat}}} = 0.6 \times 10 \times 9 \times 9 \text{ Cal} = 486 \text{ Cal}$   
 $500 \text{ cal (1 sig. fig.)}$

11.  $60 \text{ mg vitC} = 1 \text{ day}$  ;  $70 \text{ mg vitC} = 100 \text{ g orange}$  ;  $3 \text{ oz} = 1 \text{ orange}$   
 $1 \text{ week} \times \frac{7 \text{ days}}{1 \text{ week}} \times \frac{60 \text{ mg vitC}}{1 \text{ day}} \times \frac{100 \text{ g orange}}{70 \text{ mg vitC}} \times \frac{16 \text{ oz}}{454 \text{ g}} \times \frac{1 \text{ orange}}{3 \text{ oz}} =$   
 $\frac{7 \times 60 \times 100 \times 16}{70 \times 454 \times 3} \text{ oranges} = 7 \text{ oranges (1 sig. fig.)}$

12.  $5 \text{ mg tar} = 1 \text{ cig}$  ;  $0.4 \text{ mg nic} = 1 \text{ cig.}$  ;  $20 \text{ cig.} = 1 \text{ pk}$   
 $80 \cancel{\text{g tar}} \times \frac{28.35 \cancel{\text{g}}}{1 \cancel{\text{oz}}} \times \frac{1 \cancel{\text{mg}}}{10^{-3} \cancel{\text{g}}} \times \frac{1 \text{ cig}}{5 \cancel{\text{mg tar}}} \times \frac{1 \text{ pk}}{20 \cancel{\text{cig}}} =$   
 $2,000 \text{ pk (1 sig. fig.)}$

# Dimensional Analysis Key (cont)

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12. cont. How many packets of cig. = 1g nicotine

$$1 \frac{\text{g}}{\text{nic}} \times \frac{1 \text{ cig}}{0.4 \text{ mg nic}} \times \frac{1 \text{ mg}}{10^{-3} \text{ g}} \times \frac{1 \text{ pk}}{20 \text{ cigs}} =$$

$$\frac{1}{0.4 \times 10^{-3} \times 20} \text{ pk} = 125 \text{ pk} \rightarrow 100 \text{ pk (1 sig. fig.)}$$

13. 60 mi/hr. What dist. equals 1 sec.

1 sec  $\rightarrow$  dist (ft).

$$1 \frac{\text{sec}}{60 \text{ sec}} \times \frac{1 \text{ hr}}{60 \text{ min}} \times \frac{1 \text{ hr}}{1 \text{ hr}} \times \frac{60 \text{ mi}}{1 \text{ mi}} \times \frac{5280 \text{ ft}}{1 \text{ mi}} =$$

$$\frac{60 \times 5280}{60 \times 60} \text{ ft} = 88 \text{ ft} \rightarrow 90 \text{ ft (1 sig. fig.)}$$