

TRUE or FALSE

- 1) _____ Science attempts to organize and explain, in a systematic and logical manner, knowledge about our physical surroundings including our universe, our earth and ourselves.
- 2) _____ Boundaries between the scientific disciplines, like chemistry, biology and physics are very rigid.
- 3) _____ Analyze and organize data to make a general statement that explains an observation is what a theory is.
- 4) _____ The base metric quantity for mass has the symbol "m" and is called a meter.

Classify each of the following as an EXACT (E) or MEASURED (M) number..

- 5) 7 railroad cars _____
- 6) 14 gallons of gasoline _____

7) If the accepted value for the length of an object is 6.78 cm, which of the following sets of experimental results is best described as both **PRECISE AND ACCURATE**? *Circle the correct answer.*

- A) 6.88 cm, 6.88 cm, 6.58 cm (average = 6.78 cm)
- B) 6.79 cm, 6.78 cm, 6.77 cm (average = 6.78 cm)
- C) 6.71 cm, 6.71 cm, 6.72 cm (average = 6.71 cm)
- D) 6.78 cm, 6.38 cm, 6.48 cm (average = 6.55 cm)

8) Which one of the following numbers contains 4 significant figures? *MARK all that apply.*

- A) 0.0257 B) 3.090 C) 39.40 D) 9200 E) 6.43

Write the **UNCERTAIN DIGIT** and the **UNCERTAINTY (±)** in each of the following measurements.

- 9) 30 oz _____
- 10) 3.275 mL _____

11) The kelvin, which has an abbreviation of _____, is the accepted SI base unit for _____.

12) Complete the following relationships between units.

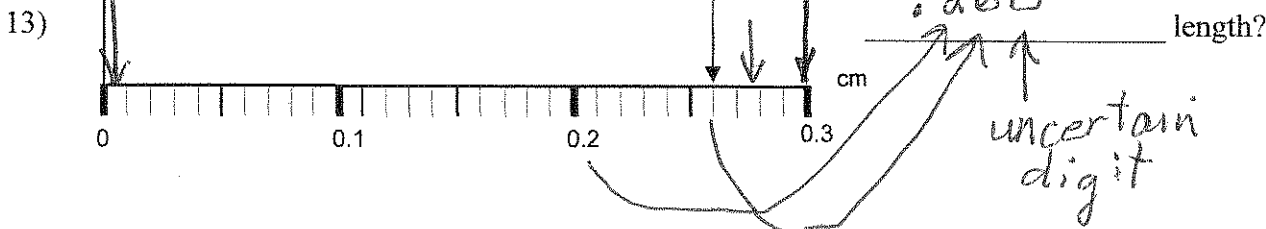
a. 10^9 g = 1 Gg

d. 1 mL = 1 cm³

b. 10^{-6} L = 1 μL

e. 10^3 Mg = 1 t (t = metric ton)

Write the value and the uncertainty (±) of the following measurements.



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8) Which one of the following numbers contains 4 significant figures? MARK all that apply.

- X A) 0.0257
- B) 3.090
- C) 39.40
- D) 9200
- E) 6.43

Write the UNCERTAIN DIGIT and the UNCERTAINTY (\pm) in each of the following measurements.

X 9) 30 oz 3, ± 10

10) 3,275 mL 5, ± 1

11) The kelvin, which has an abbreviation of K, is the accepted SI base unit for temp.

12) Complete the following relationships between units.

a. 10^9 g = 1 Gg

d. 1 mL = 1 cm

b. 10^{-6} L = 1 μ L

e. 1 Mg = 1 t (t = metric ton)

Write the value and the uncertainty (\pm) of the following measurements.

