120 pt

Mark the answers for Questions 1-34 on your Scantron. Each Question is worth 2 pt

Chp 10.1 Amino Acids

1. Which statement about amino acids is NOT correct?

- A) Naturally occurring amino acids are 'L' amino acids
- B) Amino acids in proteins are known as α -amino acids
- C) There are 20 common amino acids found in proteins.
- D) All are correct statements concerning amino acids.

2. In which of the following is an amine functional group present?

A) $CH_2=CH-CNHCH_3$ B) $(CH_3)_3NH^+$ C) NH_3 D) NH_4^+ E) CH_3NH_2 \parallel O

- 3. Which of the following is true of most amino acids found in proteins? *Mark all that are correct.* a. all of them are D-amino acids
 - b. all of them are L-amino acids
 - c. all of them are α amino acids
 - d. all of them are chiral
 - e. none of the above is true
- 4. Which amino acid has hydrogen bond attractive force associated with its side chain?

a. cysteine b. glutamine c. methionine d. glycine e. tryptophan

5. Which of these fictitious amino acids does not have a chiral center?



Use the information on the right to answer the following question.

11. Which food combination will give a complete protein?	AMINO ACID DEFICIENCY IN SOME FOODS
A) Peas and corn	RiceLys
B) Beans and Almonds	OatmealLys
C) Rice and Oatmeal	PeasMet
D) Peas and Beans	BeansTrp,Met
E) Corn and Rice	AlmondsLys, Trp
AB) Rice and Almonds	CornLys, Trp

- 12. Which of these foods is missing an essential amino acid?
 - A) Oatmeal-----Lys
 - B) Peas-----Met
 - C) Beans-----Trp, Met
 - D) Almonds-----Lys, Trp
 - E) All of them are missing an essential amino acid.

Chp 10.2 Peptide bonds

- 13. What two functional groups are involved in formation of a peptide bond?
- A) carboxylic acid and alcohol B) carboxylic acid and amine C) amide and ester D) amide and amine
- 14. The peptide bond in proteins is chemically the same as an______bond.A) esterB) alkene bondC) etherD) amideE) disulfide

(10 pt) Draw the structure of two dipeptides from one Lys and one Met.

Chp 10.3 Protein structure

15. Which of the following *is not true* about proteins

- A. Protein primary structure is in part determined by hydrogen bonding between the N-H and C=O bonds along the protein backbone.
- B. They can be denatured by alcohol.
- C. The pleated sheet structure is an example of protein secondary structure.
- D. Structural support is one function of proteins.

16. What kind of structure is shown by the diagram at the right?

- A) Primary
- B) Alpha helix
- C) Beta pleated sheet
- D) Tertiary
- E) Quaternary





22. Which of the following intermolecular forces found in the tertiary structure of proteins is the strongest? a. London b. hydrophobic c. hydrogen bonding d. a disulfide bond e. hydrophilic

Use the following peptide to answer the following questions.



- A) (2 pt) Outline the backbone of the polypeptide.
- B) (2 pt) Label the N-terminal end and the C terminal end
- C) (6 pt) <u>Draw arrows</u> showing the location of each peptide bond.
- D) (2 pt) How many amino acid residues are there?_____
- E) (6 pt) Write the primary structure of this peptide using three letter symbols for the amino acids.
- F) (6 pt) Show the hydrogen bonding that occurs in an alpha helix and label the donor and acceptor atoms.
- G) (10 pt) Write the name of the tertiary attractive force next to the side chain of each amino acid residue.

Chp 10.4 Denaturation

- 23. Acids denature proteins by disrupting
 - A) salt bridges.
 - B) hydrophobic bonds and hydrogen bonds.
 - C) peptide bonds and hydrophobic bonds.
 - D) disulfide bonds.
 - E) hydrogen bonds and disulfide bonds.

(6 pt) Enter your answers in the boxes below.



24. Which of the following will denature a protein A) heat B) stirring C) changing pH

D) adding alcohol E) all will denature proteins

(6 pt) Complete the alkaline hydrolysis reaction of the following dipeptide by drawing the structures of the products as they would occur at high pH.



Arrange the amino acids into the correct sequence using the information below.

Move the fragme show the comple	ents until they ete original de	/ are aligned to ecapeptide.	o Qu	estion 1 of 10
Amino Acids:	Ala 1 Asp 2	Gln 1 His 1	lle 1 Leu 1	Phe 2 Tyr 1
Terminal Residues:	Ala			Leu
chymotrypsin:	Ala-Tyr			
	Asp-Ile-H	lis-Phe		
	Gln-Phe			
	Asp-Leu			
pepsin:	Tyr-Asp-I	le-His		
	Phe-G1n			
	Phe-Asp-L	.eu		
thermolysin:	Ala-Tyr-A	isp		
	???-???-?	???-???-???	-???-???-	???-???-???

(5 pt) SEQUENCE:_

Chp 10.5 Protein Function

- 25. Membrane proteins can be located
 - A) entirely within the membrane
 - B) on the surface of the membrane
 - C) a) or b) above
 - D) the position cannot be determined
- 26. Which of the following is not a function of proteins?
 - A) provides structure
 - B) provide amino acids and nitrogen
 - C) provide energy
 - D) messenger molecules
 - E) part of membrane structure
 - AB) transport molecule

27. Which of the following is a function of collagen?

- A) provides structure
- B) provide amino acids and nitrogen
- C) provide energy
- D) messenger molecules
- E) part of membrane structure
- AB) transport molecule

Chp 10.6 Enzymes

- 28. What is the primary function of enzymes?
 - A) Neurotransmitters B) Biochemical catalysts

C) Structure

D) Transport

- 29. In an enzyme-substrate reaction, when excess substrate is present, increasing the concentration of the enzyme will A) decrease the turnover rate for the substrate.
 - B) increase the decomposition rate of the enzyme-substrate complex.
 - C) inhibit the formation of products.
 - D) increase the number of substrate molecules available.
 - E) increase the amount of reaction occurring.

Chp 10.7 Factors Affecting Enzyme Activity

- 30. Most enzymes are deactivated permanently above a temperature of about
 - A) 45°F. B) 25°C. C) 40°C. D) 50°C. E) 37°F.
- 31. "Physiological pH", the pH for optimum activity for most enzymes, is a pH equal to

A) 7.4. B) 8.6. C) 5.4. D) 9.0. E) 3.0.

- 32. The area on the enzyme that interacts with the substrate is called the:A) regulatory siteB) modulator siteC) active siteD) allosteric site
- 33. A molecule that is similar in structure to the substrate of an enzyme will probably be a:A) cofactorB) regulatorC) competitive inhibitorD) noncompetitive inhibitor
- 34. The pH of the environment in which an enzyme is located can influence its reactivity because a change in pH: A) can hydrolyze the protein
 - B) can produce protonation or deprotonation of amino acid side chains in the active site
 - C) changes the primary structure

D) affects the optical activity